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# Size dependence of dislocation activities and independence on theoretical elastic strain limit in Pt nanocrystals revealed by atomic-resolution *in situ* investigation



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### L. Wang <sup>a</sup>, J. Teng <sup>b</sup>, Y. Wu <sup>b</sup>, J. Zou <sup>c</sup>, G. Yu <sup>b</sup>, Z. Zhang <sup>a, d, \*\*</sup>, X. Han <sup>a, \*</sup>

<sup>a</sup> Institute of Microstructure and Property of Advanced Materials, Beijing Key Lab of Microstructure and Property of Advanced Materials, Beijing University of Technology, Beijing, 100124, China

<sup>b</sup> Department of Material Physics and Chemistry, University of Science and Technology Beijing, Beijing 100083, China

<sup>c</sup> Materials Engineering, Centre for Microscopy and Microanalysis, The University of Queensland, Brisbane, QLD 4072, Australia

<sup>d</sup> Department of Materials Science, Zhejiang University, Hangzhou, 310008, China

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#### ABSTRACT

Because of the lower total number and density of defects in nanocrystals than those in their bulk counterparts, the elastic strain limits and the plastic deformation behaviors of the former can be very different from those of the latter. Furthermore, as the surface atomic ratio increases, a surface-dominant elastic and plastic deformation characteristic may appear in nanocrystal metals. The competition between nano-strengthening and surface effects thus determines the apparent mechanical behaviors of nanocrystal metals. In this study, we conducted a series of in situ atomic-resolution deformation experiments on high stacking fault energy platinum nanocrystals using an aberration-corrected highresolution transmission electron microscope. From the direct in situ atomic-scale observations, we provided direct atomic-resolution plastic deformation mechanisms for the Pt nanocrystals of size ranging from 20 to ~0.7 nm. As the nanocrystal size decreased, a crossover occurred from dislocation slip-to dislocation-free-mediated plastic deformation. For nanocrystals of size above ~6 nm, the plastic deformation was dominated by full dislocation. However, for nanocrystals of diameters below ~2 nm, it was uncovered that the plastic deformation was dominated by the dislocation-free plastic deformation. In the elastic regime, the Pt nanocrystals reached a low elastic strain plateau by 1.5% when the size was 20 to ~9 nm. The elastic strain increased when the crystal size was below ~9 nm, and the Pt nanocrystals remained on the theoretical elastic strain limit plateau by above ~7.0% when the crystal size was below ~2 nm

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#### 1. Introduction

Small-sized single crystals usually exhibit a higher strength and sustain larger elastic strains than conventional bulk materials, because of their well-accepted size effects of defect reduction [1-4]. For these (sub) micrometer-sized face-centered cubic (FCC) single-crystal metals, it was proposed that the dislocation 'starvation' accounted for the observed ultrahigh strength [2,5-17] and the dislocation 'saturation' for the high strain plasticity [18,19]. For metals of size below ~100 nm, many experimental studies were conducted to investigate their plastic mechanism [20–25], which suggested that the deformation was dominated by partial dislocations or twining [26–32], and this was confirmed by many previous molecular dynamics (MD) simulations [33–38]. These led us to strongly believe that the plastic deformation of small-sized crystals is usually governed by the dislocation slip (full dislocation, partial dislocation or twining). It remains unclear how nanocrystals achieve plasticity and whether their size can affect the deformation behaviors when the nanocrystal size is less than 10 nm and when the surface to volume ratio is large. As suggested by previous theories and molecular simulations, when a crystal size is sufficiently small to suppress the dislocation nucleation, it is expected that elastic strain/stress can be approached, and the plasticity is controlled by breaking the atomic bonds one by one [39–43].



<sup>\*</sup> Corresponding author.

<sup>\*\*</sup> Corresponding author.

E-mail addresses: zezhang@zju.edu.cn (Z. Zhang), xdhan@bjut.edu.cn (X. Han).

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# Nanomechanical characterization of titanium incorporated gallium oxide nanocrystalline thin films

### A.K. Battu, S. Manandhar, C.V. Ramana<sup>\*</sup>

Department of Mechanical Engineering, University of Texas at El Paso, 500 W University Ave, El Paso, TX 79968, USA

#### ARTICLE INFO

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#### ABSTRACT

The effect of titanium (Ti) incorporation on the crystal structure and mechanical properties of nanocrystalline gallium oxide (Ga<sub>2</sub>O<sub>3</sub>) films (Ga-Ti-O) is reported. Ti content was varied from 0 to ~5 at% in the cosputter-deposited Ga-Ti-O films. The sputtering power applied to the Ti target was varied in the range of 0–100 W, while the sputtering power to Ga<sub>2</sub>O<sub>3</sub> was maintained at 100 W, to produce Ga-Ti-O films with variable Ti contents (0–5 at%). The Ti incorporation-induced effects were significant on the structural and mechanical properties. X-ray diffraction analysis indicated that structural transformation occurred with the increase in Ti content. The effect of Ti and associated microstructural changes are significant on the hardness (H) and elastic modulus ( $E_r$ ). The H values increased continuously from 25 to 30 GPa as a function of Ti up to 1.5 at%, after which a decreasing trend was observed. The Ga-Ti-O films exhibited excellent mechanical characteristics: H ~30 GPa,  $E_r$  ~310 GPa,  $H/E_r$  ~0.14, and  $H^3/E_r^2$  ~0.4 GPa, which are higher compared to those of intrinsic  $\beta$ -Ga<sub>2</sub>O<sub>3</sub>. On the basis of these results, a structurecomposition-mechanical property correlation in Ga-Ti-O films is established.

their properties and performance.

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#### 1. Introduction

Gallium oxide (Ga<sub>2</sub>O<sub>3</sub>)-based thin films and nanomaterials have received considerable attention in recent years because of their wide range of technological applications in photonics, electronics, optoelectronics, magneto-electronics, chemical sensing, and catalysis [1–5]. Among the known wide band gap oxides, the scientific, technological merits, and potential of Ga<sub>2</sub>O<sub>3</sub> have been widely recognized [6-8]. Ga<sub>2</sub>O<sub>3</sub> exhibits polymorphism; it can crystallize in five different crystal structures, namely  $\alpha$ ,  $\beta$ ,  $\gamma$ ,  $\delta$ , and  $\varepsilon$  phases [9,10]. Among these polymorphs, monoclinic  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> is thermodynamically favorable and stable with a wide band gap of ~4.8 eV [9,10].  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> exhibits *n*-type conductivity related to donor centers involving oxygen vacancies and/or impurities [11-13]. The intrinsic β-Ga<sub>2</sub>O<sub>3</sub> finds numerous technological applications in electronics, photonics, catalysis, electro-optics, gas sensing, deep ultraviolet (UV) photo electronics, and magnetoelectronics [14–17]. The thermal stability and higher melting point make  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> ideal for applications in extreme environmental conditions of hightemperature chemical sensors, catalysis, and thin-film transistors

\* Corresponding author. E-mail address: rvchintalapalle@utep.edu (C.V. Ramana). properties of oxides for a given technological application [21]. In fact, designing new and advanced materials must combine the excellent structural and electronic properties, which are comparable to those of currently existing materials, along with

[18–20]. However, the fundamental understanding of the physics and chemistry of Ga<sub>2</sub>O<sub>3</sub>-based nanomaterials is important for all

the aforementioned applications. Such fundamental studies and

understanding will allow the optimization of synthetic processes and conditions to provide a better control on the interplay among

surface/interface structure, thermodynamic conditions, chemical

processes, and kinetics, which in turn can facilitate the control of

selectively doping mechanically resilient Ti into Ga oxide (GTO) for

extreme environment applications, such as high-temperature

sensors, combustion systems, and photodetectors. The impetus

for the present work is to derive a fundamental understanding of

the nanomechanical behavior of GTO films with variable Ti contents

and the effect of Ti on the structural and mechanical properties.

Under reduced dimensions, especially on transition from micro- to nanoscale, the materials' performance is closely related to their

ultra-microstructure and phase composition, and it also depends

on the mechanical characteristics. Therefore, a controlled phase

and microstructure are extremely important to improve the desired

The goal of the present work is to design novel materials by







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# Ductility and plasticity of nanostructured metals: differences and issues

Y.T. Zhu <sup>a, b, \*</sup>, X.L. Wu <sup>c, d, \*\*</sup>

<sup>a</sup> Nano and Heterostructural Materials Center, Nanjing University of Science and Technology, 200 Xiaolingwei Road, Nanjing 210094, China

<sup>b</sup> Department of Materials Science and Engineering, North Carolina State University, 1001 Capability Drive, Raleigh, NC 27695, USA

<sup>c</sup> State Key Laboratory of Nonlinear Mechanics, Institute of Mechanics, Chinese Academy of Sciences, 15 Beisihuan West Road, Beijing 100190, China

<sup>d</sup> College of Engineering Sciences, University of Chinese Academy of Sciences, 19A Yuquan Road, Beijing 100049, China

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#### ABSTRACT

Ductility is one of the most important mechanical properties for metallic structural materials. It is measured as the elongation to failure of a sample during standard uniaxial tensile tests. This is problematic and often leads to gross overestimation for nanostructured metals, for which non-standard small samples are typically used. Uniform elongation is a better measure of ductility for small samples because they are less sensitive to sample size. By definition, ductility can be considered as tensile plasticity, but it is often confused with plasticity. In principle, ductility is largely governed by strain hardening rate, which is in turn significantly affected by microstructure, whereas plasticity is primarily controlled by crystal structure or the number of available slip systems to accommodate plastic deformation. In practice, ductility is important for preventing catastrophic failure of structural components during service, whereas plasticity is critical for shaping and forming metals into desired shape and geometry to make structural components. Nanostructured metals typically have high plasticity, but low ductility, due to their low strain hardening capability. Increasing strain hardening rate via modifying microstructure is the primary route to improving ductility.

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#### 1. Introduction

Reasonable ductility (usually >5%, preferably >10%) is desired to prevent mechanical components or structures from catastrophic failure during service [1]. On the other hand, high strength is also desired so that a metallic structure/component can carry large load at low material weight. This is especially important for future transportation vehicles such as electrical cars, which need to be lightweight to improve their energy efficiency. However, a metallic material is either strong or ductile, but rarely both at the same time [2,3]. Coarse-grained (CG) metals usually have high ductility but low strength. Refining grains to the nanocrystalline regime in the last few decades has significantly increased strength, but this is often accompanied with the sacrifice of ductility [4]. The low ductility of nanostructured metals has been a major issue with their potential structural applications. Ductility of nanostructured metals has been a hot research topic for over a decade [2–7]. However, despite the extensive research and publications, there still exist widespread confusions and misconceptions on the definition and measurement of ductility of nanostructured metals, which have led to the publications of problematic claims and data.

The biggest confusion is on the difference between ductility and plasticity. Plasticity is an important property for metallic materials, which could significantly affect their processing, shaping, and forming ability. Unfortunately, in the academic literature, these two terminologies are often mixed up and interchanged, which has raised serious issues and sometimes led to wrong and/or misleading scientific claims and statements. What is more problematic is that such publications often mislead the research community, especially junior researchers and students, as well as the public. To make things worse, plasticity and ductility are often not well defined in textbooks, e.g. Deformation and Fracture Mechanics of Engineering Materials [8]. These problems become more serious in recent years with the study of nanostructured metals, in which very small, non-standard samples are often used to characterize mechanical properties.

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<sup>\*</sup> Corresponding author.

<sup>\*\*</sup> Corresponding author.

E-mail addresses: ytzhu@ncsu.edu (Y.T. Zhu), xlwu@imech.ac.cn (X.L. Wu).

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# Nanospace within metal-organic frameworks for gas storage and separation

B. Li<sup>a, e</sup>, H.-M. Wen<sup>b, e</sup>, Y. Yu<sup>a</sup>, Y. Cui<sup>a</sup>, W. Zhou<sup>d</sup>, B. Chen<sup>a, c, \*</sup>, G. Qian<sup>a, \*\*</sup>

<sup>a</sup> State Key Laboratory of Silicon Materials, Cyrus Tang Center for Sensor Materials and Applications, School of Materials Science and Engineering, Zhejiang University, Hangzhou, 310027, PR China

<sup>b</sup> College of Chemical Engineering, Zhejiang University of Technology, Zhejiang, 310014, PR China

<sup>c</sup> Department of Chemistry, University of Texas at San Antonio, One UTSA Circle, San Antonio, TX 78249-0698, USA

<sup>d</sup> NIST Center for Neutron Research, National Institute of Standards and Technology, Gaithersburg, MD 20899-6102, USA

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#### ABSTRACT

Porous metal—organic frameworks (MOFs), also known as porous coordination polymers, represent a new class of porous materials, and one of their striking features lies in their tunable, designable, and functionalizable nanospace. This nanospace within MOFs provides virtually plenty of room for imagination, allowing designed incorporation of different size, shape, and functionalities for targeted gas storage and separation applications. Furthermore, the features of high porosities, tunable framework structures and pore sizes, and immobilized functional sites enable MOF materials to fully make use of their nanopore space for gas storage, to optimize their sieving effects, and to differentiate their interactions with gas molecules for gas separation. In this review article, we highlight some recent significant advances in developing microporous MOFs for some of the most important gas storage and separation applications.

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#### 1. Introduction

With the industrial revolution in the last two centuries, gas molecules have become essential chemicals and energy resources and have been widely used in our daily life. For example, hydrogen, methane (the main component of shale and natural gases), and acetylene can provide large amounts of energy through their combustion oxidation. Light hydrocarbons are very important energy resources and raw chemicals in the petrochemical industry. The intensive usage of petroleum-based fuels both in industry and our daily life produces tremendous amounts of carbon dioxide ( $CO_2$ ) into the atmosphere, leading to the concentration of atmospheric  $CO_2$  sharply rising from 278 ppm at the beginning of industrial revolution to >400 ppm today. These gases play important roles and influences in our daily life—an 'age of gas' is apparently dawning. However, a gas is a form that is very difficult to handle

\*\* Corresponding author.

because it is easily dispersed, creates mixtures, and is invisible. Conventional transportation, storage, and separation of gases consume massive amounts of energy and require harsh conditions and high cost (e.g. very high storage pressure and energy-intensive cryogenic distillation). It is thus highly important for scientists and engineers to develop efficient materials for diverse gas storage and separation applications.

Porous materials with nanosized pores have contributed significantly to the science and technology of gas handling. Porous materials have been around for centuries; traditional materials are activated carbon and zeolite, both of which have shown many applications in the petroleum industry, catalysis, and gas separation by making use of their nanopores. However, the nanospace of these traditional materials is quite limited in terms of tuning space size, shape, and functionalization, basically only through control of the thermal activation and substitutions of metal cations. The discovery of novel porous materials whose properties are superior to conventional materials would revolutionize gas handing technologies and thus benefit our society. In this context, the emergence of porous metal-organic frameworks (MOFs) has attracted extensive attentions from both academia and industry due to their tunable, designable, and functionalizable nanospace [1–5]. MOF materials can be straightforwardly self-assembled through the coordination





 $<sup>\</sup>ast$  Corresponding author. Department of Chemistry, University of Texas at San Antonio, USA.

*E-mail addresses*: banglin.chen@utsa.edu (B. Chen), gdqian@zju.edu.cn (G. Qian).

<sup>&</sup>lt;sup>e</sup> These authors have contributed equally to this work.

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## Indocyanine green—encapsulated nanoscale metal—organic frameworks for highly effective chemo-photothermal combination cancer therapy

K. Jiang <sup>a, c</sup>, L. Zhang <sup>a, c</sup>, Q. Hu <sup>b</sup>, D. Yue <sup>a</sup>, J. Zhang <sup>a</sup>, X. Zhang <sup>a</sup>, B. Li <sup>a, \*</sup>, Y. Cui <sup>a</sup>, Y. Yang <sup>a</sup>, G. Qian <sup>a, \*\*</sup>

<sup>a</sup> State Key Laboratory of Silicon Materials, Cyrus Tang Center for Sensor Materials and Applications, School of Materials Science and Engineering, Zhejiang University, Hangzhou, 340027 China

<sup>b</sup> Department of Pharmacology, School of Medicine, Hangzhou Normal University, Hangzhou 310036 China

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#### ABSTRACT

Indocyanine green (ICG), as the only U.S. Food and Drug Administration—approved near-infrared (NIR) clinical agent, has been considered as an ideal light absorber for laser-mediated photothermal therapy (PTT) in cancer treatment. However, the practical applications of ICG are severely hampered by its poor aqueous stability, rapid body clearance, and low cellular uptake. To overcome these limitations, we herein report the successful example of integrating ICG into a zeolitic imidazolate framework (ZIF-8) to fabricate a novel nanoscale ICG@ZIF-8 hybrid material. Through a simple one-pot synthesis method, a high loading content of 20.6% can be achieved in the resultant ICG@ZIF-8. The photostability and tumor accumulation of ICG are notably promoted due to the protection of the framework, leading to enhanced photothermal conversion efficiency. Furthermore, we also discover, for the first time, that the pH-triggered release of large amount of  $Zn^{2+}$  from ZIF-8 in tumor acidic microenvironment also significantly contributes to targeted killing of cancer cells. As a result of the combined PTT and chemotherapy, ICG@ZIF-8 exhibits greatly improved diagnostic efficacy for both *in vitro* and *in vivo* cancer therapy, leading to 91% tumor eradication in all the mice treated with ICG@ZIF-8 and NIR irradiation. Hematoxylin and eosin (H&E)—stained slices show that no noticeable tissue damage is observed in major organs, indicating the safety of ICG@ZIF-8.

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#### 1. Introduction

Photothermal therapy (PTT), which uses an optical absorbing agent to efficiently convert light energy into heat which causes a rise in the local temperature beyond  $42^{\circ}$ C to consequently kill cancer cells, has been widely recognized as a promising non-invasive strategy for future cancer treatment. Much effect has been dedicated to the development of near-infrared (NIR) laser (ranging from 700 to 1,100 nm)–assisted PTT because of its low tissue adsorption, deep penetration ability, and low phototoxicity [1–3]. Until now, a large number of NIR-adsorbing inorganic agents,

\*\* Corresponding author.

including gold nanostructures [4–7], carbon nanotubes [8–11], and copper sulfide nanoparticles (NPs) [12], have been extensively used as PTT agents. However, these inorganic agents are nonbiodegradable and generally have potential long-term toxicity, which seriously limit their clinical applications. In this regard, indocyanine green (ICG), the only U.S. Food and Drug Administration—approved NIR agent, exhibits higher photothermal conversion efficiency and better biocompatibility [13]. However, ICG is not an effective PTT agent for practical applications because of its easy photodegradation, rapid blood clearance (an ultra-short half-life period of 2–4 min), and low tumor accumulation rate [14–16]. Therefore, the development of novel platforms to improve the photostability and tumor accumulation of ICG is urgently desirable for highly efficient PTT.

To overcome the aforementioned issues, various NP-based delivery systems, such as polyallylamine, perfluorocarbon, polypeptide micelles, and poly (lactic-co-glycolic acid)–constituted





<sup>\*</sup> Corresponding author.

E-mail addresses: bin.li@zju.edu.cn (B. Li), gdqian@zju.edu.cn (G. Qian).

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# Advances in understanding atomic-scale deformation of small-sized face-centered cubic metals with *in situ* transmission electron microscopy

### X. Wang, L. Zhong<sup>\*\*</sup>, S.X. Mao<sup>\*</sup>

Department of Mechanical Engineering and Materials Science, University of Pittsburgh, Pittsburgh, PA, 15261, United States

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#### ABSTRACT

Face-centered cubic (fcc) metals exhibit outstanding mechanical properties both in small-sized and bulk nanostructures and are thus promising for applications in micro/nano electromechanical systems (M/ NEMS). Uncovering the intrinsic deformation mechanisms of small-sized metals is of critical importance for evaluating their feasibility in M/NEMS applications. Recent advances in *in situ* transmission electron microscopy (TEM)—based mechanical testing techniques open up opportunities for achieving a mechanistic understanding of the atomic-scale deformation mechanisms. This article reviews recent progresses in *in situ* TEM studies on the various plastic deformation modes of small-sized fcc metals, including dislocation slip and twinning, phase transformation—mediated plasticity, reversible structure formation, diffusion-mediated plasticity, and void-assisted plasticity and fracture. Promising directions for future *in situ* TEM investigations on fcc metals are also enumerated.

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#### 1. Introduction

The excellent mechanical performances of nanostructured materials generally originate from their unique deformation mechanisms [1,2]. Deformation modes of materials are known to be size and structure dependent [3,4]. Revealing the competition and interaction between different deformation modes is technically demanding yet critical for the design and process of advanced materials. Applications in the micro/nano electromechanical systems (M/NEMS) necessitate interrogating the mechanical properties of nanostructured materials at small scale (e.g. <50 nm in diameter) [5], which further increases the technical difficulties involved therein. To date, the understanding of atomistic deformation mechanisms has been highly dependent on computational simulations [6–9], the reliability of which can be affected by their inherent high strain rates (i.e.  $10^{7-9}$ /s in computation vs.  $<10^{-1}$ /s in laboratory) and accuracy of the applied interatomic potentials. As a result, advanced experimental techniques capable of direct observation on the dynamic deformation process are highly appealing.

\*\* Corresponding author.

E-mail addresses: liz50@pitt.edu (L. Zhong), sxm2@pitt.edu (S.X. Mao).









<sup>\*</sup> Corresponding author.